

α-Alkenoyl Ketene S,S- and N,S-Acetals As Starting For Unexpected and Novel Synthesis of N-Heterocycles

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ABSTRACTr

A series of *N*-heterocycles such as diazaspiro-[4.5]decane, pyrazolo[4,3-d][1,2]-diazepine, imidazo[3,2,1ij][1,8]naphthyridine derivatives or pyrazolo[3,4-b]pyridin-4-ol were synthesized using α -alkenoyl-, α , α -dialkenoyl ketene-(S,S)acetals **2a-d**, **3a-c** or α -dialkenoyl ketene-(*N*,*S*)-acetal **12** as starting materials. The biological activity of some new synthesized compounds have been investigated.

Indexing terms/Keywords

Oxo-ketene Acetals; α -Alkenoyl ketene -S,S-acetal; α -Alkenoyl keten-N,S-acetal; Diazaspiro[4.5]decane Pyrazolodiazepine; Pyrazolopyridine

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INTRODUCTION

Indeed, the first synthesis of an α -oxo ketene-(*S*,*S*)-acetal was reported in 1910 by Kelber and co-workers [1]. This compound was prepared by alkylation of β -oxodithioic acid which was obtained in poor yield by reaction of an aryl ketone with CS₂ and KOH at 100 °C followed by neutralization with sulfuric acid. In the early 1960s Thuillier and Vialle successfully prepared α -oxo ketene-(*S*,*S*)-acetals directly from ketones in good yields by using sodium tert-amylate as the base and two equivalents of an alkyl halide [2,3]. Subsequent workers devoted their efforts towards improving the synthetic methodology and achieved mild reaction conditions, high yields and wide generality to wide substrates for the synthesis of α -oxo ketene-(*S*,*S*)-acetals [3]. However, there are only a few reports so far on the synthesis of α -oxo ketene-(*S*,*S*)-acetals [4] and none in water.

The use of water as a solvent in organic chemistry was rediscovered in the 1980s in Breslow's work, which showed that hydrophobic effect could strongly enhance the rates of some organic reactions [5]. Organic reactions carried out in water, without the use of any organic solvent, can also be beneficial because water is an easily available, cheap, safe and environmentally benign solvent [6]. A clean, facile and practical synthesis of α -oxo ketene-(*S*,*S*)-acetals based on the reaction of α -dicarbonyl compounds with carbon disulfide and alkyl bromide catalyzed by tetrabutyl ammonium bromide (TBAB) in the presence of potassium carbonate in water [7,8].

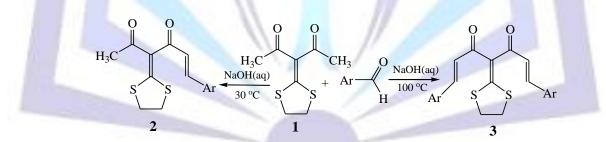
El-Saghier et.al, [9] reported that the synthesis 2-acetyl-2-oxopropylidene-(S,S)-acetal was obtained via reaction of acetyl acetone, CS_2 and two moles of methyl iodide in one-pot reaction using phase transfer Conditions PTC [K_2CO_3 /benzene/tetrabutyl ammonium bromide (TBAB)] in almost 100% yield.

Liu *et. al.*[10], they reported the important synthesis of novel α -alkenoyl ketene-(*S*,*S*)-acetals in aqueous media as intermediate in organic synthesis. They investigated the aldol condensation reactions of α , α -diacetyl ketene-(*S*,*S*)-acetals with various selected aryl aldehydes affording a variety of novel α -alkenoyl ketene-(*S*,*S*)-acetals in water.

The important synthetic utility of such intermediate [11-17] showed that α -alkenoyl ketene-(*S*,*S*)-acetals containing a dienone moiety showed promising structural features as novel intermediates for: (1) double Michael acceptors serving as five carbon 1,5-bielectrophilic species, (2) dense and flexible substitution patterns and (3) good leaving alkyl thio groups that can be subjected to a nucleophilic vinyl substitution (SNV) reaction. Consequently, they developed a novel synthetic strategy for the construction of highly substituted six-membered carbocycles and heterocycles, relying upon the utilization of α -alkenoyl ketene-*S*,*S*-acetals as a five carbon 1,5-bielectrophilic species informal [5+1] annulation with various carbon, nitrogen and sulphur nucleophiles, respectively [18].

Results and Discussions

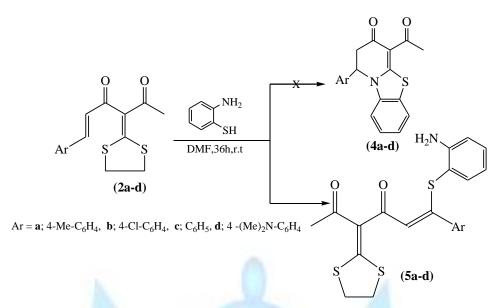
The synthesis and application of \Box -oxo ketene-(*S*,*S*)-acetals as starting material in heterocyclic synthesis have been reported in the recent work in our laboratory [9,22-29] and elsewhere.[1-4, 20,21] Following the procedure described by Liu *et. al* [10], α -alkenoyl- and α , α \Box -dialkenoyl ketene-(*S*,*S*)-acetals **2**, **3** were easily prepared via direct aldol condensation reactions of some α , α ²-diacetyl ketene-(*S*,*S*)-acetals **1** with aromatic aldehydes.



In the present work, we found that the use of α -alkenoyl ketene-(*S*,*S*)-acetals **2** or α , α '-dialkenoyl ketene-(*S*,*S*)-acetals **3** as intermediate in heterocylic synthesis afforded an expected and unexpected novel *N*-hetercycles.

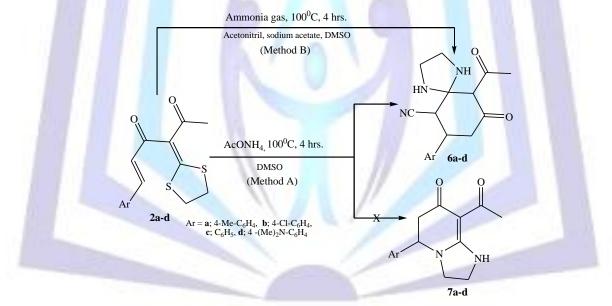
A solution of 2-(propan-2-ylidene)-1,3-dithiolane **2a-d** were allowed to react with 2-aminothiophenol in DMF in one portion at room temperature afforded (Z)-6-(2-aminophenythio)-3-(1,3-dithiolan--2-ylidene)-6-arylhex-5-ene-2,4-dione **5a-d** instead of the expected 3-acetyl-5,6-dihydro-6-arylbenzthiazolo[3,2-a]-pyridine-4-one **4a-d**.





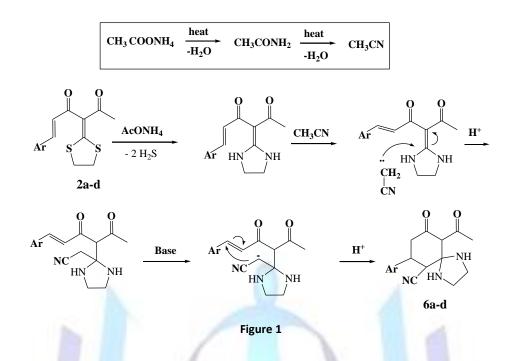
The spectral data of the synthesized compounds are in accordance with the proposed structures. The IR (v cm⁻¹) spectra of compounds **4a-d** showed new bands at 3370-3280 cm⁻¹ for NH₂ groups; and the ¹H-NMR spectra showed a new singlet at 4.60-4.40 ppm for (NH₂).

Unexpected synthesis of 1,4-diazaspiro[4.5]decane-6-carbonirile derivatives **6a-d** were obtained via reaction of 3-(1,3-dithiolan-2-ylidene)-6-aryl-hex-5-ene-2,4-dione **2a-d** with ammonium acetate in DMSO under heating at 100 ⁰C instead of 8-acetyl-2,3,5,6-tetrahydro-5-arylimidazo-[1,2-a]pyridin-7(1H)-one **7a-d**.



The reaction mechanism was assumed to proceed via a nucleophilic attack of ammonia (which produced from decomposition of AcONH₄) at the ethylenic bond of ketene-(*S*,*S*)-acetals followed by elimination of H₂S to give the corresponding ketene-(*N*,*N*)-acetals. Under the higher temperature, the ammonium acetate is converted into acetonitrile by lost of two moles of water (Figure 1). Under the reaction condition, the carbanion of acetonitrile which was generated under basic condition attack at the ethylenic bond of ketene-(*N*,*N*)-acetals followed by another nucleophilic attack of the newly formed carbanion at the ethylenic bond of α , β -unsaturated alkenoyl and cyclization.



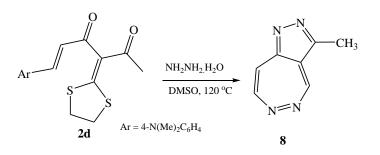


Entry Comp. No		Conditions	Yield %	
1	6a	AcONH₄/ DMSO/4hrs./100 ⁰ C	75	
2	6b	AcONH₄/ DMSO/4hrs./100 ^⁰ C	84	
3	6c	AcONH ₄ / DMSO/4hrs./100 ⁰ C	58	
4	6d	AcONH ₄ / DMSO/4hrs./100 ⁰ C	63	
5	6a	NH _{3(g)} , CH ₃ CN, AcONa [*]	69	
6	6b	NH _{3(g)} , CH ₃ CN, AcONa [*]	77	
7	6c	NH _{3(g)} , CH ₃ CN, AcONa [*]	55	
8	6d	NH _{3(g)} , CH ₃ CN, AcONa [*]	59	

^{*} Method (B)

This mechanism was proved by the synthesis of compounds **2a-d** via reaction of the starting materials **2a-d** with acetonitrile and ammonia gas at 100° C in DMSO for about 4 hrs. in the presence of sodium acetate as a base (method B entry 5-8). The yield of the reaction is moderately high in an entry (1-4) than in an entry (5-8) where compound **6b** was 84 % in entry 3 but was 77% in entry 6 (Table 1).

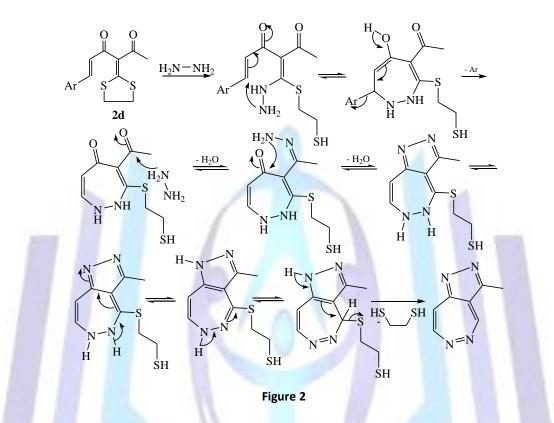
Also, the unexpected (3aZ,5Z,7Z)-3-methylpyrazolo[4,3-d][1,2]-diazepine **8** was prepared via reaction of (*E*)-3-(1,3-dithiolan-2-ylidene)-6-(4-(*N*,*N*-dimethylamino)phenylhex-5-ene-2,4-dione **2d** with hydrazine hydrate was carried out in DMSO at 120 °C in an oil bath with stirring for about 4hrs.





The ¹H-NMR spectra showed the disappearance of the aromatic and CH_2 signals and appearance of ethylic CH of diazepines ring. The ¹³C-NMR spectra showed the disappearance carbonyl group signals and the appearance the signals correspondence to the compound **8**.

The reaction mechanism is assumed to proceed via a nucleophilic attack of two moles of hydrazine, the first attack at the ethylenic double bond of ketene-(S,S)-acetals followed by cyclization to diazepine ring with elimination of aromatic ring, and the second mole attack at carbonyl carbon with elimination of two moles of moles of water followed by elimination of dithioethanol to afford the desired compound **8** (c.f. Figure 2).



On the other hand, compound **2a** reacted with phenyl hydrazine at 120 °C in DMSO to afford the corresponding the imidazodiazepinone **9**.

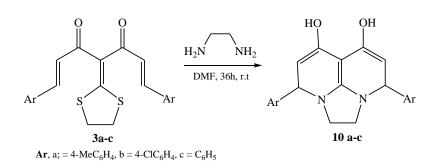


The IR spectrum data for compound **9** showed a new band at 3194 cm-1 for NH group. The ¹H-NMR showed a signal at 7.70 ppm as a singlet for 1H (NH); and 7.60-6.80 ppm as a multiplet for aromatic protons; 3.40 ppm for (<u>CH₂-CO</u>). The ¹³C-NMR) spectrum showed the appeared in opposite direction when applying DEPT technique; a new signal at 31.12 ppm for C7 of diazepin ring.

On the other hand, way, α, α' -dialkenoyl ketene-(*S*,*S*)-acetals **3a-c** were allowed to react with ethylenediamine in one pot reaction at room temperature , afforded the corresponding 1,2,4,9-tetrahydro-4,9-diarylylimidazo[3,2,1ij][1,8]-naphthyrid-ine-6,7-diol derivatives **10a-c**.

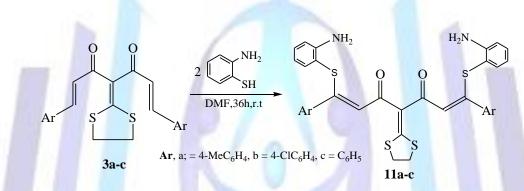
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The spectral data of the synthesized compounds **10a-c** are in accordance with the proposed structures. The IR v cm⁻¹ spectra for **10a-c** showed new bands for OH group at 3286-3282 cm⁻¹ and the ¹H-NMR spectra showed a new signals at 9.60-9.50 ppm as singlet for OH group; a doublet at 3.80-3.70 ppm for (2H, 2CH-Ar) and appeared as revered to opposite direction when applying DEPT technique; a singlet at 2.01 ppm for (2CH₃) of compound **10a**.

Also, **3a-c** were allowed to react with 2 moles of 2-aminothiophenol under the same condition and reaction mechanism of compound **5** to afford (1Z,6Z)-1,7-bis(2-aminophenylthio)-4-(1,3-dithiolan-2-ylidene)-1,7-diarylhepta-1,6-diene-3,5-dione **11a-c**



In the same strategy, reaction of \Box -dialkenoyl ketene-(*N*,*S*)-acetal **12** with hydrazine hydrate was heated at 120 °C in DMSO for 2 hrs gave 3-methyl-6-*p*-tolyl-6H-pyrazolo[3,4-b]pyridin-4-ol **13**



The spectrum date of compound 13 is accordance with the proposed structures.

The suggested reaction mechanism may include two steps. The first involves nucleophilic attack of amino group of hydrazine at the ethylenic bond with elimination of MeSH followed by nucleophilic attack of the other amino group at the carbonyl group followed by cyclization to pyrazole ring. The second step involved a nucleophilic attack of the amino group of a second mole of hydrazine at \Box \Box \Box -unsaturated bond followed by a nucleophilic attack of the amino group at the ethyleinc bond with elimination of aniline followed by elimination of NH₃.

Biological Activity Assessment.

The biological activity of the compounds **(5b-d, 11b,c and 10b,c)** given in Tables **(2, and 3)** regard less their relatedness and nuclei were assessed using the paper disc assay procedure adapted by *Baur et al.*[28] (1966). This method was minorly modified and adapted to cover all compounds under investigation.

Compounds were dissolved in the suitable organic solvents to make stock solutions of final concentrations of 1000 and down to 100 μ g /20 μ L. Paper disks of 8 mm diameter (Whatmann No. 1) are loaded with 20 ml of each individual compound under testing, solvent was evaporated under a stream of hot air.

Using sterile forceps discs were aseptically applied onto the surface of nutrient-agar plates (12 cm/diameter) previously inoculated with 0.25 ml of 24 hrs culture of the test organism (two Gram-positive or two Gram-negative species). Plates



were then left for 1 hr at 4 $^{\circ}$ C for better diffusion of the loaded test compounds, and then transferred to an incubator to grow for 24 hrs at 37 $^{\circ}$ C. At the end of incubation period the diameters of clear inhabitation zones (Diz_s) were measured in triplicates.

Negative control discs loaded with 20 ml of the solvents were tested alongside the compounds while positive control discs containing specific affective concentrations of strptamycin, Methicillin and Erythromycin were tested in parallel. Results are recorded and expressed as mean of triplicate sets and DIZ were approximated to the nearest integrated value.

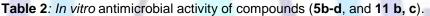
Test Organisms

For testing the antimicrobial activity of the compounds under study, four test bacterial species, namely *Bacillus subtilis*, *Staphylococcus aureus* (G +ve), *scherichia coli* and *Klebsiella pneumonia* (Gram –ve) were used. All the above mentioned organisms were kindly gifted forms Clinical Microbiology Laboratory, Tripoli Medical Centre, Tripoli, Libya, while antimicrobial activity assessments were carried out by the technical Staff in Biolab, Chemistry Department, Faculty of Science, Tripoli University, and Tripoli, Libya.

Results and Discussion

2-Aminothiophenol Derivatives

The in vitro antibacterial activity of 2-aminothiophenol derivatives presented in (Table 2) using a limited spectrum of microbes belong to Gram-positive and Gram-negative species. The results demonstrated that, among 5 tested individual 2-aminothiophenol, compounds (5b, c, d, and 11 b, c) displayed their highest activity against B. subtilis (D1Z ~ 20-22 mm) at 500 μ g/disc. On the other hand compound No. (11c) showed mild anti G-positive potential while all tested compounds were completely inactive against both Gram-negative tested bacteria.



Diameter of Inhibition Zone (mm)				µg\Disk	Solvent	Compd No.
Gram+ve		Gram-ve		1		
B.sub	St.aur	E.coli	K.pne		. /	
-ve	-ve	15	20	500	Acetone	5b
-ve	-ve	15	21	500	Acetone	5c
-ve	-ve	15	20	500	Acetone	5d
-ve	-ve	18	22	500	Acetone	11b
-ve	-ve	15	17	500	Acetone	11c
-ve	-ve	-ve	-ve		Acetone	Control
9	9	20	20	10	Disc	Streptomycin*

B.sub; Bacillus subtilis St.aur; Staphylococcus aureus; E.coli; Escherichia coli K.pne Klebsiella pneumonia, * Acts on DNA

Diaza bicyclo[7.2.1]dodeca derivatives 10b,c.

 Table 3: In vitro antimicrobial activity of compounds (10 b, c):

Diameter of Inhibition Zone (mm)				µg\Disk	Solvent	Compd No.
Gram+ve		Gram-ve				
B.sub	St.aur	E.coli	K.pne			
-ve	-ve	11	12	500	Acetone	10b
-ve	-ve	11	14	500	Acetone	10c
-ve	-ve	-ve	-ve		Acetone	control
9	9	20	20	10	Disc	Streptomycin*
-ve	-ve	-ve	-ve	5	Disc	Methicillin**
-ve	-ve	-ve	-ve	15	Disc	Erythromycin***

B.sub Bacillus subtilis ;

St.aur Staphylococcus aureus ; E.coli Escherichia coli K.pne Klebsiella pneumonia.



* Acts on DNA ** Acts on cell membrane *** Acts on protein synthesis.

Acknowledgments

We are grateful to Libyan Petroleum Institute, International Center for Polymer Research, Central Laboratory, Cairo University, Egypt) for recording IR,¹H NMR,¹³C NMR and Mass spectra and Prof. Abdel-Kader N. for biological activity Assessment.

Experimental section

All melting points were determined on a Koffler melting point apparatus and are uncorrected.¹H NMR spectra were recorded on a Bruker avance 300 MHz spectrometer using TMS as an internal reference (chemical shifts in \Box , ppm),¹³C NMR spectra were recorded on a Bruker avance 75 MHz spectrometer using TMS as an internal reference (chemical shifts in \Box , ppm), IR in KBr were obtained on a Bruker FTIR ISS 25 spectrophotometer (mmax in cm⁻¹) and The Mass spectra were recorded on Shimadzu GCMS-QP 1000 EX (Japan) mass spectrometer at 70 eV.

(Z)-6-(2-aminophenythio)-3-(1,3-dithiolan--2-ylidene)-6-arylhex-5-ene-2,4-dione (5a-d).

General Procedure

A solution of α -alkenoyl ketene-(*S*,*S*)-acetals **2a-d** (1.0 mmol) in DMF (5 ml) was added 2-aminothiophenole (3.0 mmol) in one portion at room temperature. The reaction mixture was stirred for 36 hrs at room temperature and then poured into saturated sodium chloride aqueous (25 ml). This was washed with water, filtered. The crude product was purified by petroleum ether to give the desired title compounds **5a-d**.

(Z)-6-(2-aminophenylthio)-3-(1,3-dithiolan-2-ylidene)-6-p-tolylhex-5-ene-2,4-dione 5a

(83 %), m.p. 101-102 °C [Found: C, 61.82; H, 5.01; N, 3.17; O, 7.50; S, 22.48, $C_{22}H_{21}NO_2S_3$ (427.6) requires C, 61.79; H, 4.95; N, 3.28; O, 7.48; S, 22.50; IR (v cm⁻¹) 3369, 3289 (NH₂), 3061 (CH_{arom.}), 2939 (CH_{aliph.}), 1695, 1614 (2C=O). cm⁻¹; (¹H-NMR, 300 MHz CDC1₃, δ ppm) 7.70-6.70 (m, 8H, arom.), 6.90 (s, 1H, CH-S), 4.43 (s, 2H, NH₂), 3.42 (m, 4H, 2CH₂), 2.40 (s, 3H, CH₃-Ar), 2.30 (s, 3H, CH₃-CO).,¹³C-NMR 75.47 MHz, CDCl₃) 193.46 (<u>C</u>O), 191.92 (<u>C</u>O), 172. 64 (<u>C</u>=S(S)), 148.62 (=<u>C</u>(S, Ar), 145.18, 141.50, 136.81, 131.69, 129.78, 128.95, 127.72, 125.36, 118.21 (<u>C_{arom.}</u>), 118.70 (=<u>C</u>(CO,CO)) 15.23 (=<u>C</u>H), 37.81, 36.79 (2<u>C</u>H₂), 29.49 (CH₃-Ar), 21.56 (<u>C</u>H₃-CO).

(Z)-6-(2-aminophenylthio)-6-(4-chlorophenyl)-3-(1,3-dithiolan-2-ylidene)hex-5-ene-2,4-dione 5b

 $(88\%), m.p. 104-108 \ ^{o}C [Found: C, 56.30; H, 4.05; Cl, 7.91; N, 3.13; O, 7.14; S, 21.47, C_{21}H_{18}CINO_2S_3 (448.02) requires C, 56.35; H, 3.98; Cl, 7.99; N, 3.11; O, 7.23; S, 21.40,; IR (v cm⁻¹) 3370, 3280 (NH₂), 3006 (CH_{arom}), 2930 (CH_{aliph}), 1674, 1620 (2C=O).). cm⁻¹; (¹H-NMR, 300 MHz ,CDC1₃, <math>\bar{o}$ ppm) 7.80-6.90 (m, 8H, arom.), 7.30 (s, 1H, CH), 4.50 (s, 2H, NH₂), 3.47 (m, 4H, 2CH₂), 2.50 (s, 3H, CH₃-CO).¹³C-NMR 75.47 MHz, CDCl₃) 194.06 (<u>C</u>O), 192.02 (<u>C</u>O), 173. 04 (<u>C</u>=S(S)), 149.52 (=<u>C</u>(S, Ar), 145.38, 142.20, 137.21, 131.71, 129.80, 128.20, 127.75, 125.39, 118.24 (<u>C</u>arom.), 118.74 (=<u>C</u>(CO,CO)) 15.23 (=<u>C</u>H), 38.31, 38.12 (2<u>C</u>H₂), 21.57 (<u>C</u>H₃-CO).

(Z)-6-(2-aminophenylthio)-3-(1,3-dithiolan-2-ylidene)-6-phenylhex-5-ene-2,4-dione 5c

(66 %), m.p. 87-90 °C [Found: C, 60.99; H, 4.63; N, 3.39; O, 7.74; S, 23.26, $C_{21}H_{19}NO_2S_3$ (413.58) requires C, 60.89; H, 4.91; N, 3.32; O, 7.52; S, 22.48; IR (v cm⁻¹) 3360, 3298 (NH₂), 2983 (CH_{arom.}), 2930 (CH_{aliph.}), 1700, 1665 (2C=O). cm⁻¹; (¹H-NMR, 300 MHz CDC1₃, δ ppm) 7.74-6.76 (m, 9H, arom.), 6.92(s, 1H, CH), 4.56 (s, 2H, NH₂), 3.54-3.34 (m, 2H, 2CH₂), 2.40 (s, CH₃-CO)..;¹³C-NMR 75.47 MHz, CDCI₃) 193.16 (<u>C</u>O), 191.82 (<u>C</u>O), 173. 64 (<u>C</u>=S(S)), 149.12 (=<u>C</u>(S, Ar), 145.22, 141.53, 136.82, 132.19, 129. 87, 128. 59, 127.62, 125.46, 118.23 (<u>C</u>arom.), 118.70 (=<u>C</u>(CO,CO)), 115.24 (=<u>C</u>H), 37.81, 37.79 (2<u>C</u>H₂), 21.56 (<u>C</u>H₃-CO).

(Z)-6-(2-aminophenylthio)-6-(4-(dimethylamino)phenyl)-3-(1,3-dithiolan-2-ylidene)hex-5-ene-2,4-dione 5d

(87%), m.p. 87-88 °C [Found: C, 60.49; H, 5.30; N, 6.13; O, 7.01; S, 21.07, $C_{23}H_{24}N_2O_2S_3$ (456.64) requires C, 60.54; H, 5.35; N, 6.10; O, 6.98; S, 21.00; IR (v cm⁻¹) 3369, 3289 (NH₂), 3061 (CH_{arom}), 2939 (CH_{aliph}), 1695, 1614 (2C=O) cm⁻¹; (¹H-NMR, 300 MHz CDC1₃, δ ppm) 7.70-6.70 (m, 8H, arom.), 6.90 (s, 1H, CH), 4.50 (s, 2H, NH₂), 3.40 (m, 2H, 2CH₂), 2.40 (s, 6H, N (CH₃)₂, 2.30 (s, 3H, CH₃-CO); ¹³C-NMR 75.47 MHz, CDCl₃) 193.46 (<u>C</u>O), 191.92 (<u>C</u>O), 172. 64 (<u>C</u>=S(S)), 148.62 (=<u>C</u>(S, Ar), 145.18, 141.50, 136.81, 131.69, 129.78, 128.95, 127.72, 125.36, 118.21 (<u>C</u>_{arom}.), 118.70 (=<u>C</u>(CO,CO)) 115.23 (=<u>C</u>H), 37.81, 36.79 (2<u>C</u>H₂), 29.49 (<u>C</u>H₃)2N), 21.56 (<u>C</u>H₃-CO).

2-Acetyl-6-cyano-5-arylcyclohexspiro(1,2')-imidazolidine-3-one **6a-d**

General Procedure

To a solution of α -alkenoyl ketene-*S*,*S*-acetal **2a-d** (1 mmol) in DMSO (5 ml) was added ammonium acetat (2 mmol). The reaction mixture was heated to 100 °C in oil bath under stirring for 4 hrs. After cooling to room temperature, the



mixtures were poured into saturated sodium chloride solution (15 ml). The combined filtered, and. the crude products **6a-d** were recrystalized from ethanol.

2-Acetyl-6-cyano-5-(p-tolyl)-cyclohex spiro(1,2')-imidazolidine-3-one 6a.

(75 %), m.p. 218-220 °C [Found: C, 69.43; H, 6.80; N, 13.49; O, 10.28 $C_{18}H_{21}N_3O_2$ (311.16) requires C, 69.27; H, 6.79; N, 13.38; O, 10.38; IR (v cm⁻¹) 3235 (NH), 3067 (CH_{arom}), 2955, 2912 (CH_{aliph}), 2206 (CN) 1627, 1573 (2C=O) cm⁻¹; (¹H-NMR, 300 MHz CDC1₃, $\bar{0}$ ppm) 7.20 (s, 4H, arom.), 6.40 (s, 1H, NH), 4.60 (m, 1H, CH-Ar), 3.07 (s, 1H, CO-CH-CO), 2.60 (d, 1H, CH-CN), 2.44 (s, 3H, CH₃-CO), 2.35 (s, 3H, CH₃-Ar), 1.70 (m, 4H, 2CH₂).; ¹³C-NMR 75.47 MHz, CDCl₃) 195.76(CO), 187.52 (CO), 167.88 (<u>C</u>(NH,NH)), 139.28, 134.86, 130.02, 126.55 (C_{arom}), 116.37 (<u>C</u>N), 86.33 (<u>C</u>(CO,CO)), 57.04 (<u>C</u>H(CN)), 42.12, 37.14 (<u>C</u>H₂-<u>C</u>H₂), 30.27 (CH₂-CO), 29.17 (<u>C</u>H-Ar), 21.17 (<u>C</u>H₃-CO), 20.94 (<u>C</u>H₃-Ar).

2-Acetyl-6-cyano-5-(p-chlorophenyl)-cyclohexspiro(1,2')-imidazolidine-3-one 6b.

(84 %) , m.p. 119-122 $^{\circ}$ C [Found: C, 61.54; H, 5.47; Cl, 10.69; N, 12.66; O, 9.64 C₁₇H₁₈Cl N₃O₂ (331.8) requires C, 61.60; Cl, 10.71; H, 5.45; N, 12.68; O, 9.68; ; IR (v cm⁻¹) 3206 (NH), 3067 (CH_{arom}), 2913 (CH_{aliph}), 2207 (CN), 1610, 1600 (C=O) cm⁻¹; (¹H-NMR, 300 MHz CDC1₃, δ ppm) 7.30 (m, 4H, arom.), 6.02 (s, 1H, NH), 4.61 (m, 1H, CH-Ar), 3.20 (s, 1H, CO-CH-CO), 2.61 (d, 1H, CH-CN), 2.43 (s, 3H, CH₃-CO), 1.72 (m, 4H, 2CH₂).;¹³C-NMR 75.47 MHz, CDCl₃) 195.77 (CO), 187.55 (CO), 167.87 (<u>C</u>(NH,NH)), 139.25, 135.06, 130.12, 127.15, 126.25 (C_{arom}.), 116.30 (<u>C</u>N), 86.13 (<u>C</u>(CO,CO)), 57.06 (<u>C</u>H(CN)), 42.15, 37.20 (<u>C</u>H₂-<u>C</u>H₂), 30.30 (CH₂-CO), 29.20 (<u>C</u>H-Ar), 21.28 (<u>C</u>H₃-CO).

2-Acetyl-6-cyano-5-phenyl-cyclohexspiro (1,2')-imidazolidine-3-one 6c.

(58 %), m.p. 125-127 °C [FoundC, 68.67; H, 6.44; N, 14.13; O, 10.76 $C_{17}H_{19}N_3O_2$ (297.35) requires C, 68.73; H, 6.45; N, 14.22; O, 10,78; IR (v cm⁻¹) 3200 (NH), 3058 (CH_{arom}), 2912, 2905 (CH_{aliph}), 2206 (CN), 1627, 1600 (C=O) cm⁻¹; (¹H-NMR, 300 MHz CDC1₃, δ ppm) 7.35 (s, 5H, arom.), 6.42 (s, 1H, NH), 4.63 (m, 1H, CH-Ar), 3.17 (s, 1H, CO-CH-CO), 2.59 (d, 1H, CH-CN), 2.43 (s, 3H, CH₃-CO), 1.71 (m, 4H, 2CH₂); ¹³C-NMR 75.47 MHz, CDCl₃) 196.57 (CO), 187.73 (CO), 168.78 (<u>C</u>(NH,NH)), 139.25, 134.06, 131.02, 127.15 (C_{arom}), 116.50 (<u>C</u>N), 87.13 (<u>C</u>(CO,CO)), 57.34 (<u>C</u>H(CN)), 43.12, 38.14 (<u>C</u>H₂-<u>C</u>H₂), 30.20 (CH₂-CO), 29.23 (<u>C</u>H-Ar), 21.28 (<u>C</u>H₃-CO).

2-Acetyl-6-cyano-5-(p-(N,N-dimethylamino) phenyl)-cyclohexSpiro(1,2')-imidazolidine-3-one 6d).

($63 \ \%$), m.p. $185-188^{\circ}$ C [Found: C, 67.04; H, 7.11; N, 16.46; O, 9.40, $C_{19}H_{24}N_4O$ (340.32) requires C, 67.11; H, 7.18; N, 16.39; O, 9.48; IR (v cm⁻¹) 3233 (NH), 3073 (CH_{arom.}), 2984, 2816 (CH_{aliph}), 2208 (CN), 1650, 1610 (C=O) cm⁻¹; (¹H-NMR, 300 MHz CDC1₃, δ ppm) 7.26 (s, 4H, arom.), 6.40 (s, 1H, NH), 4.61 (m, 1H, CH-Ar), 3.07 (s, 1H, CO-CH-CO), 2.63 (d, 1H, CH-CN), 2.44 (s, 6H, N (CH₃)₂), 2.35 (s, 3H, CH₃-CO), 1.70 (m, 4H, 2CH₂); ¹³C-NMR 75.47 MHz, CDCl₃) 195.77 (CO), 187.53 (CO), 167.85 (<u>C</u>(NH,NH)), 139.25, 135.06, 130.12, 127.15 (C_{arom.}), 116.30 (<u>C</u>N), 86.13 (<u>C</u>(CO,CO)), 57.04 (<u>C</u>H(CN)), 42.10, 37.10 (<u>C</u>H₂-<u>C</u>H₂), 30.30 (CH₂-CO), 29.21 (<u>C</u>H-Ar), 21.28 (<u>C</u>H₃-CO), 21.22 (N(<u>C</u>H₃)₃).

(3aZ,5Z,7Z)-3-Methylpyrazolo[4,3-d][1,2]-diazepine 8

To a solution of 6-(4-(N,N-dimethylamino) phenyl)-3-(1,3-dithiolan-2-ylidine)-hex-5-ene-2,4-dione (2d) (1 mmol) in DMSO (5 ml) was added hydrazine (2 mmol). The reaction mixture was heated to 120 °C in oil bath under stirring for 4 hrs. After cooling to room temperature, the mixture was poured into sodium chloride (15 ml). The combined filtered, and. The crude product was purified by chloroform and petroleum ether (2:2), gave compound **8**, (65.60 %), m.p. 253-255 °C [Found: C, 57.53; H, 4.14; N, 38.34; C₇H₆N₄ (148.16) requires C, 57.59; H, 4.15; N, 38.28; IR (v cm⁻¹) 3086 (CH _{arom}), 2911, 2803 (CH _{aliph}). cm⁻¹; (¹H-NMR, 300 MHz CDC1₃, δ ppm) 8.60 (s, 1H, CH-N), 7.80 (d, 1H, CH-N), 6.70 (d, 1H, CH-C=N), 3.10 (s, 3H, CH₃); ¹³C-NMR 75.47 MHz, CDCl₃) 160.78 (C₅, C=N), 152.12 (C₄ _{pyrazole ring}), 129.19 (C₆, diazepine ring), 122.11 (C=N (CH₃), CH), 111.72 (C₅, diazepine ring), 76.60 (C₃, diazepine ring), 40.20 (CH₃).

9-Acetyl-5-phenyl-1-(phenylamino)-6-p-tolyl-2,3,6,7-tetrahydro-1H-imidazo[1,2-b][1,2]diazepin-8(5H)-one 9.

To a solution of 3-(1,3-dithiolan-2-ylidine)-6-*p*-tolylhex-5-ene-2,4-dione **2a** (1 mmol) in DMSO (5 ml), phenylhydrazine (2 mmol) was added. The reaction mixture was heated to 120 °C in oil bath under stirring for 2:30 hrs. After cooling to room temperature, the mixture was poured into sodium chloride (15 ml). The combined filtered, and. The crude product was purified by chloroform and petroleum ether, gave compound **9**, 92.3 %) , m.p. 88-90 °C [Found C,72.33; H, 6.49; N, 14.54; O, 6.64, $C_{29}H_{31}N_5O_2$ (481.59) requires C, 72.29; H, 6.45; N, 14.58; O, 6.68; IR (v cm⁻¹) 3194 (NH), 2913(CH_{arom}), 2819(CH_{aliph}), 1625, 1603(2C=O) cm⁻¹; (¹H-NMR, 300 MHz CDC1₃, δ ppm) 7.70 (s, 1H, NH), 7.60-6.80 (m, 14H, arom.), 3.35 (m, 6H, 2CH₂, CH₂), 3.10 (m, 1H, CH-Ar), 2.39 (s, 3H, CH₃-Ar), 2.30 (s, 3H, CH₃-CO).; ¹³C-NMR 75.47 MHz, CDCl₃) 195.96 (CO), 193.0 (CO), 172.67 (<u>C</u>(CO,CO)), 146.34, 145.15, 144.12, 143.25, 141.49, 138.36, 131.68, 129.54, 129.43, 129.21, 128.81, 126.74, 125.86, 123.14 (C_{arom}), 37.80, 37.50, 37.13 (3CH₂), 31.12 (<u>C</u>H-Ar), 21.56 (<u>C</u>H₃-CO).

4,9-Diaryl-1,2,4,9-tetrahydroimidazo[3,2,1ij][1,8]naphthyridine-6,7-diol **10a-c.**

General Procedure

To a solution of α, α -dialkenoyl ketene-(*S*,*S*)-acetals **3a-c** (1.0 mmol) in DMF (5 ml) ethylenediamine (1.2 mmol) was added in one portion at room temperature. The reaction mixture was stirred for 48 hrs at room temperature and then



poured into saturated sodium chloride aqueous (25 ml). This was washed with water, filtered. The crude product was purified by ethanol+H₂O (2:2), gave product **10a-c**.

4,9-Di-p-tolyl-1,2,4,9-tetrahydroimidazo[3,2,1-ij][1,8] naphth-yridine-6,7-diol 10a.

(85 %), m.p. 192-196°C [Found: C, 77.39; H, 6.49; N, 7.52; O, 8.59, $C_{24}H_{24}N_2O_2$ (372.46) requires C, 77.39; H, 6.53; N, 7.58; O, 8.58; IR (v cm⁻¹) 3286 (OH), 3020 (CH _{arom}), 2918 (CH _{aliph})cm⁻¹; (¹H-NMR, 300 MHz CDC1₃, \bar{o} ppm) 9.50 (s, 1H, OH), 7.60-7.09 (m, 8H, arom.), 6.90 (d, 2H, 2CH=C-OH), 3.70 (d, 2CH, 2CH-Ar), 2.40 (s, 4H, 2CH₂), 2.30 (s, 6H, 2CH₃); ¹³C-NMR 75.47 MHz, CDCl₃) 188.68 (<u>C</u>(OH)), 166.71 (<u>C</u>(N,N)), 138.64 (<u>C</u>-Ar), 139.74, 132.87, 129.88, 128.46, (C_{arom}), 101.74 (<u>C</u>(C-OH,C-OH)), 77.90 (CH_{Pyridine}, CH), 46.64 (2<u>C</u>H₂), 21.15 (<u>C</u>H₃).

4,9-Bis(4-chlorophenyl)-1,2,4,9-tetrahydroimidazo[3,2,1-ij]-[1,8]naphthyridine-6,7-diol 10 b.

(83 %), m.p. 134-138 °C [Found: C, 63.93; H, 4.39; Cl, 17.16; N, 6.78; O, 7.74, $C_{22}H_{18}$ Cl₂N₂O₂ (413.3) requires C, 64.11; H, 4.35; N, 6.68; O, 7.78; IR (v cm⁻¹) 3286 (OH),3020 (CH _{arom.}),2918 (CH _{aliph.}) cm⁻¹; (¹H-NMR, 300 MHz CDC1₃, δ ppm) 9.60 (s, 1H, OH), 7.60-7.09 (m, 8H, arom.), 6.80 (d, 2H, 2CH=C-OH), 3.80 (d, 2CH, 2CH-Ar), 2.40 (s, 4H, 2CH₂); ¹³C-NMR 75.47 MHz, CDCl₃) 188.78 (<u>C</u>(OH)), 167.51 (<u>C</u>(N,N)), 139.14 (<u>C</u>-Ar), 138.94, 133.17, 129.85, 128.46 (C_{arom.}), 102.04 (<u>C</u>(C-OH,C-OH)),), 77.86 (CH_{Pyridine}, CH), 47.64 (2<u>C</u>H₂).

4,9-Diphenyl-1,2,4,9-tetrahydroimidazo[3,2,1-ij][1,8] naphth-yridine-6,7-diol 10c.

(72 %), m.p. 120-122 $^{\circ}$ C [Found: C, 76.72; H, 5.85; N, 8.13; O, 9.29, $C_{22}H_{20}N_2O_2$ (344.41) requires C, 76.79; H, 5.90; N, 8.18; O, 9.28; IR (v cm⁻¹) 3282 (OH), 3020 (CH arom.), 2918 (CH aliph.) cm⁻¹; (¹H-NMR, 300 MHz CDC1₃, δ ppm) 9.60 (s, 1H, OH), 7.70-7.23 (m, 9H, arom.), 6.89 (d, 2H, 2CH=C-OH), 3.70 (d, 2CH, 2CH-Ar), 2.50 (s, 4H, 2CH₂); ¹³C-NMR 75.47 MHz, CDCl₃) 188.77 (<u>C</u>(OH)), 166.73 (<u>C</u>(N,N)), 139.14 (<u>C</u>-Ar), 138.94, 133.17, 129.45, 129.85, 128.46 (C_{arom.}), 100.16 (<u>C</u>(C-OH,C-OH)),), 77.93 (CH_{Pyridine}, CH), 47.54 (<u>2CH₂</u>).

1,7-Bis(2-aminophenylthio)-4-(1,3-dithiolane-2-ylidine)-1,7-diary1hepta-1,6-diene-3,5-dione 11a-c

General Procedure

To a solution of α, α -dialkenoyl ketene-(*S*,*S*)-acetal **3a-c** (1.0 mmol) in DMF (5 ml) 2-aminothiophenol (3.0 mmol) was added in one portion at room temperature. The reaction mixture was stirred for 36 hrs at room temperature and then poured into saturated sodium chloride aqueous (25 ml). This was washed with water, filtered. The crude products were purified by ethanol+H₂O (1:1), gave products **11a-c**.

1,7-Bis(2-aminophenylthio)-4-(1,3-dithiolane-2-ylidine)-1,7-di-p-toly1hepta-1,6-diene-3,5-dione 11a.

(70 %), m.p. 100-104 $^{\circ}$ C [Found: C, 66.22; H, 4.94; N, 4.29; O, 4.90; S, 19.64, $C_{36}H_{32}N_2O_2S_4$ (652.91) requires C, 66.30; H, 4.95; N, 4.28; O, 4.88; IR (v cm⁻¹) 3372, 3292 (NH₂), 3061(CH_{arom}), 2919 (CH_{aliph}), 1621 (CO).cm⁻¹; (¹H-NMR, 300 MHz CDC1₃, δ ppm) 7.50-6.50 (m, 18H, 16H arom +2H of 2CH), 4.10 (br, 4H, 2NH₂), 3.40 (s, 4H, 2CH₂), 2.53 (s, 6H, 2CH₃).;¹³C-NMR 75.47 MHz, CDCI₃) 187.86 (<u>C</u>O), 148.64 (<u>C</u>=S(S)), 148.59 (=<u>C</u>(S, Ar), 143.83, 136.74, 134.61, 131.62, 130.55, 128.40, 128. 06, 127. 80, 125.77, 118.22, 115.24 (<u>C</u>_{arom}), 118.74 (=<u>C</u>(CO,CO)), 76.70 (=<u>C</u>H), 37.33 (<u>C</u>H₂),), 22.17 (*C*H₃-Ar).

1,7-Bis(2-aminophenylthio)-1,7-bis(4-chlorophenyl)-4-(1,3-dithiolane-2-ylidine)-hepta-1,6-diene-3,5-dione **11b**.

 $(64 \%), \text{m.p. } 89-90 \ ^{\circ}\text{C} [Found: C, 58.86; H, 3.78; Cl, 10.22; N, 4.04; O, 4.61; S, 18.49, C_{34}H_{26}Cl_2N_2O_2S_4 (693.75) \ \text{requires} C, 58.79; H, 3.75; Cl, 10.11; N, 4.08; O, 4.63; S, 18.50; IR (v cm^{-1}) 3372, 3294 (NH_2), 3063, 3022 (CH_{arom.}), 2925 (CH_{aliph}), 1623 (CO) cm^{-1}; (^{1}\text{H-NMR}, 300 \ \text{MHz} \ \text{CDC1}_3, \delta \ \text{ppm}) \ 7.70-6.50 \ \text{(m, 18H, 16H arom +2H of 2CH}), 3.80 \ \text{(br, 4H, 2NH}_2), 3.35 \ \text{(s, 4H, 2CH}_2); (^{13}\text{C-NMR} \ 75.47 \ \text{MHz}, \ \text{CDCl}_3) \ 187.89 \ (\underline{C}O), 148.65 \ (\underline{C}=S(S)), 148.60 \ (=\underline{C}(S, Ar), 143.80, 136.84, 134.71, 131.62, 130.53, 128.42, 128. 03, 127. 87, 125.87, 118.24, 115.23 \ (\underline{C}_{arom.}), 118.73 \ (=\underline{C}(CO,CO)), \ 76.66 \ (=\underline{C}H), 37.35 \ (\underline{C}H_2). \$

1,7-Bis(2-aminophenylthio)-4-(1,3-dithiolane-2-ylidine)-1,7-diphenylhepta-1,6-dione-3,5-diene 11c.

(81 %), m.p. 109-111 °C [Found: C, 65.35; H, 4.52; N, 4.48; O, 5.12; S, 20.53, C34H28N2O2S4 (624.86) requires C, 65.39; H, 4.55; N, 4.45; O, 5.18; S, 22.50; IR (v cm⁻¹) 3372, 3294 (NH₂), 3063, 3022 (CH_{arom}), 2925 (CH_{aliph}), 1624 (CO) cm⁻¹; (¹H-NMR, 300 MHz CDC1₃, δ ppm) 7.80-6.50 (m, 20H, 18H arom +2H of 2CH), 4.40 (br, 4H, 2NH₂), 3.40 (s, 4H, 2CH₂); ¹³C-NMR 75.47 MHz, CDCI₃) 187.83 (<u>C</u>O), 148.62 (<u>C</u>=S(S)), 148.60 (=<u>C</u>(S, Ar), 143.81, 136.84, 134.71, 131.62, 130.53, 128.42, 128. 03, 127. 87,125.87, 118.24, 115.23 (<u>C</u>arom.), 118.72 (=<u>C</u>(CO,CO)), 76.60 (=<u>C</u>H), 37.30 (<u>C</u>H₂),

3-Methyl-6-p-tolyl-6H-pyrazolo[3,4-b]pyridin-4-ol 13.

A solution of 3,3-(methylthiophenylamino)methylene)-6-*p*-tolylhexa-5-ene-2,4-dione **(12)** (1 mmol) in DMSO (5 ml) hydrazine (1 mmol) was added. The reaction mixture was heated to 120 °C in oil bath under stirring for 2 hrs. After cooling to room temperature, the mixture was poured into sodium chloride (15 ml). The combined filtered, and the crude product was purified by CHCl₃+ pet. Ether (1:3), gave compound **13**, (60.38 %), m.p. 242-243^d °C [Found: C, 70.28; H, 5.48; N, 17.56; O, 6.69, C₁₄H₁₃N₃O (239.27) requires C, 70.29; H, 5.45; N, 17.58; O, 6.68; IR (v cm⁻¹) 3322 (OH), 3095 (CH_{arom.}), 2932 (CH_{aliph}) cm⁻¹; (¹H-NMR, 300 MHz CDC1₃, δ ppm) 9.40 (s, 1H, OH), 7.50-7.30 (m, 4H, arom.), 6.90 (d, 1H, CH=C), 2.61 (s, 1H, CH) 2.40 (s, 3H, CH₃-Ar), 2.53 (s, 3H, CH₃CO);¹³C-NMR 75.47 MHz, CDCl₃) 193.78 (C-OH), 154.00



 $(\underline{C}=(N),(N)), 141.0 \ (C_{Bridge}), 129.19-117.97 \ (\underline{C}_{arom.}), 121.0 \ (\underline{C}H-Ar), 106.16 \ (\underline{C}-CH_3), 76.60 \ (\underline{C}H_{Pyridine}) 29.51 \ (\underline{C}H_3-Ar), 14.11 \ (\underline{C}H_3).$

REFERENCES

- [1] Kelber C. Ber. Dtsch. Chem. Ges.; 1910, 43, 1252
- [2] a). Thuillier A., Vialle J.; Bull. Soc. Chim. Fr. 1962, 2187-2193; b)

Thuillier A., Vialle J.; Bull. Soc. Chim. Fr.; 1962, 2194 –2198.

- [3] a). Larsson F.; Lawesson S.; *Tetrahedron.;* 1972,28, 5341–5347; b). Corey E. J.R. Chen H. K.; *Tetrahedron Lett.;* 1973, 14, 3817–3820; c). Dieter R. K.; *J. Org.Chem.*; 1981, 46, 5031–5033.
- [4] a). Dalgaard L., Kolind-Andersen H., Lawesson S.; *Tetrahedron.;* 1973, 29,2077–2085;b). Dalgaard L., Jensen L., Lawesson S.O.; *Tetrahedron.;* 1974, 30, 93 –104.
- [5] a). Breslow R., Rideout D. C.; J. Am. Chem. Soc.; 1980, 102,7816 7817; b). Breslow R.; Acc. Chem. Res.;1991,24, 159 – 164.
- [6] a). Aqueous-Phase Organometallic Catalysis. Conceptsand Applications ; Eds.: (Cornils, B. Herrmann W. A.), Wiley-VCH, Weinheim ,1998; b). Organic Synthesis in Water, (Ed.:P. A. Grieco), Blackie Academic and Professional:London, 1998.; c). Li C., Chan T., Organic Reactionsin Aqueous Media ; John Wiley & Sons, NewYork,1997; d). Kobayashi S., Manabe K.; In Stimulating Concepts in Chemistry.; (Eds.: Shibasaki M., Stoddart J., Vogtle F.).; Wiley-VCH, Weinheim.; 2000.
- [7] a). Li C.; Chem. Rev.; 1993, 93, 2023–2035.; b).Genet J.; Savignac M.; J. Organomet. Chem.; 1999, 576, 305 317;c). Fringuelli F., Piermatti O., Pizzo F., Vaccaro L.; Eur. J.Org. Chem.; 2001, 439 –452; d). Kobayashi S., Manabe,K.; Acc. Chem. Res.; 2002, 35, 209 217.
- [8] a). Kobayashi S.; Wakabayashi T.; *Tetrahedron Lett.*; **1998**,39, 5389 –5392.; b). Manabe K., Mori Y., Kobayashi S.; *Synlett.*; **1999**, 1401–1402.; c). Manabe K., Kobayashi S.; *Org. Lett.*; **1999**, 1, 1965–1967; d). Otto S., Engberts J., Kwak J.; *J. Am. Chem. Soc.*; **1998**, 120, 9517 –9525.; e). Rispens T., Engberts J.; *Org. Lett.* **2001**, **3**, 941–943.; f). Manabe K., Sun X., Kobayashi S.; *J.Am. Chem. Soc.*; **2001**, 123, 10101 10102.
- [9] El_Shafei A., El-Saghier A. M, Ahmed E.; Synthesis; 1994, 2,152-154.
- [10] Ouyang Y., Dong D., Pan W., Zhang J. and Liu Q., J. Tetrahedron.; 2006, 62,10111-10116.
- [11] Huang Z., Liu Z.; Synthesis.; 1978, 357
- [12] Gompper R., Toep ft.W.; cheme .Ber.; 1962, 95, 2861.
- [13] Hiria K., Matsuda H., Kishda Y.; chem. Pharm. Bull.; 1971, 20,97.
- [14] Mansour N., Rudor W., Augustian M.; Chem.; 1981, 21, 69.
- [15] Rajappa S., Advani B.; Proc. Indian Acad. Sci. (chem. Sci.).; 982, 91, 463.
- [16] Huang Z., Shi X.; Synthesis.; 1990,162.
- [17] a). Dieter R. K.; *Tetrahedron.*; **1986**, 42, 3029–3096.;b). Tominaga Y.; *J. Heterocycl. Chem.*; **1989**, 26, 1167 1204.;c). Junjappa H., Ila H., Asokan C. V.; *Tetrahedron.*; **1990**, 46, 5423 5506;d). Kolb M.; *Synthesis.*; **1990**, 171–190.; e).Junjappa H., Ila .H.; *Phosphorus, Sulfur Silicon.*;**1994**, 35, 95 96; f)., Junjappa H., Ila. H., Mohanta P.; in: *Progress in Heterocyclic Chemistry.*;(Eds.:Gribble G., Gilchrist L.); Pergamon Press, Oxford.; **2001**, (13), Ch.1, 1 24.
- [18] a). Singh G., Purkayastha M., Ila H., Junjappa H.; *J. Chem. Soc.* Perkin Trans .1; 1985, 1289–1294.; b).
 Kocienski P., Pontiroli J., Liu Q.; *J. Chem. Soc. Perkin Trans.*1; 2001, 2356–2366; c). Asokan C., Junjappa V., Ila H.; Synthesis. ; 1987, 284 285
- [19] a). Apparao S., Datta A., Ila H.; Synthesis.; 1985, 169 171; b). Datta A., Ila H., Junjappa H.; Tetrahedron Lett.; 1988, 29, 497–500.
- [20] a). Balu M., Ila H., Junjappa H.; Tetrahedron Lett.; 1987, 28, 3023 3026.; b). Dieter R., Fishpaugh J.; J. Org. Chem.; 1988, 53, 2031–2046.
- [21] a). Dieter R., Silks L., Fishpaugh J., Kastner M.; J. Am. Chem. Soc. 1985, 107, 4679 4692.; b). Dieter R., Fishpaugh J., Silks L.; Tetrahedron Lett. 1982, 23, 3751–3754.
- [22] El-Saghier A. M. M., Alwedi E. F. and Fawzy N. M., J. Heterocyclic Chem., 2012, 49, 664.
- [23] Khodairy A. and El-Saghier A. M., Acta Chim. Solv., 2011, 58, 360-366.
- [24] Mohamed M. A. A., Fawzy N. M., and El-Saghier A. M., Chemistry J., 2012, 2(2), 101-105.
- [25] Amed E. A., Mohamed M. A. A. and El-Saghier A. M., J. Am. Sci., 2012, 8(8), 815-818



ISSN 2321-807X

- [26] Farhat M. F., Makhlouf M. A., El-Saghier A. M., Mezoughi A. B. A., Awhida S. M. and El-Mehdi A. M., *Arb. J. Chem.*, **2011**, 4, 307-3011.
- [27] For Review: El-Saghier A. M., Abdel-Ghany H, Mohamed A. A. M. and Younes S. H., Trend Org. Chem., 2011, 15, 1-24.
- [28] Bauer A. W., Kirby M. M., Sherris J. C. and Truck M., Antibiotic Susceptibility testing by a Standardized single disk method., Am. J. Clin. Pathol, 1966, 45, 493-6.

